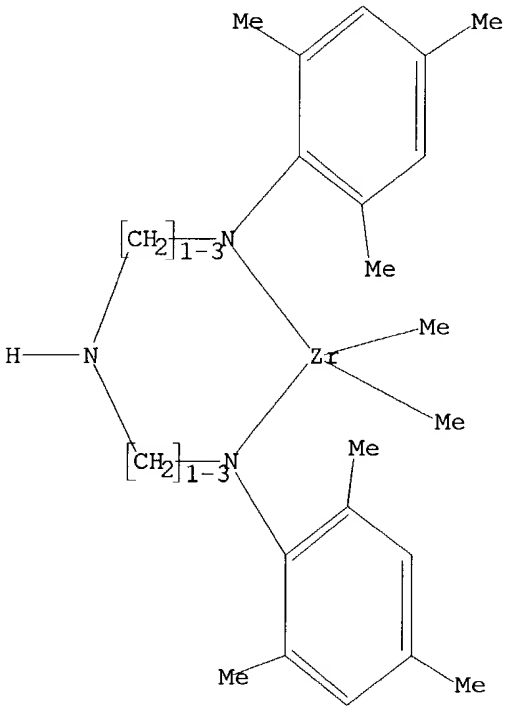
09/864,756

(FILE 'HOME' ENTERED AT 16:54:58 ON 08 JUL 2004)

FILE 'REGISTRY' ENTERED AT 16:56:18 ON 08 JUL 2004
L1 STRUCTURE UPLOADED

=> d l1
L1 HAS NO ANSWERS
L1 STR



Structure attributes must be viewed using STN Express query preparation.

=> s l1
SAMPLE SEARCH INITIATED 16:56:50 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 12 TO ITERATE

100.0% PROCESSED 12 ITERATIONS 0 ANSWERS
SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**
BATCH **COMPLETE**
PROJECTED ITERATIONS: 33 TO 447
PROJECTED ANSWERS: 0 TO 0

L2 0 SEA SSS SAM L1

=> s l1 full
FULL SEARCH INITIATED 16:56:57 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 236 TO ITERATE

100.0% PROCESSED 236 ITERATIONS 4 ANSWERS
SEARCH TIME: 00.00.01

L3 4 SEA SSS FUL L1

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=> fil caplus
COST IN U.S. DOLLARS                                    SINCE FILE      TOTAL
                                                           ENTRY    SESSION
FULL ESTIMATED COST                                       155.42     155.84

FILE 'CAPLUS' ENTERED AT 16:57:03 ON 08 JUL 2004
USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT.
PLEASE SEE "HELP USAGETERMS" FOR DETAILS.
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FILE COVERS 1907 - 8 Jul 2004  VOL 141 ISS 2
FILE LAST UPDATED: 7 Jul 2004  (20040707/ED)

  This file contains CAS Registry Numbers for easy and accurate
  substance identification.


=> s l3
L4            3 L3

=> d 1-3 bib abs

L4   ANSWER 1 OF 3  CAPLUS  COPYRIGHT 2004 ACS on STN
AN   2000:824311  CAPLUS
DN   134:5262
TI   Catalysts and method for polymerization of olefins
IN   McConville, David H.; Schrock, Richard R.
PA   Univation Technologies, LLC, USA; Massachusetts Institute of Technology
SO   PCT Int. Appl., 37 pp.
     CODEN: PIXXD2
DT   Patent
LA   English
FAN.CNT 1
     PATENT NO.         KIND  DATE          APPLICATION NO.  DATE
     ---------------    ----  --------      ---------------  --------
PI   WO 2000069922      A2    20001123      WO 2000-US13312  20000515
     WO 2000069922      A3    20010208
         W:  AU, BR, BY, CA, CN, CZ, ID, IL, IN, JP, KR, MX, NO, PL, RU, SG,
             SK, TR, ZA, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
             PT, SE
     US 6271325         B1    20010807      US 1999-312878   19990517
     EP 1185563         A2    20020313      EP 2000-932437   20000515
         R:  AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO
     TR 200103289       T2    20020321      TR 2001-20010328920000515
     BR 2000010534      A     20020416      BR 2000-10534    20000515
     JP 2002544339      T2    20021224      JP 2000-618337   20000515
     AU 756239          B2    20030109      AU 2000-50159    20000515
     TW 526207          B     20030401      TW 2000-89109277 20000515
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	EG 23015	A	20031231	EG 2000-624	20000515
	US 2001041778	A1	20011115	US 2001-864756	20010524
	ZA 2001009439	A	20030217	ZA 2001-9439	20011115
	NO 2001005613	A	20011116	NO 2001-5613	20011116
PRAI	US 1999-312878	A	19990517		
	WO 2000-US13312	W	20000515		

OS MARPAT 134:5262

AB This invention relates to a composition of matter represented by formula (I), and to a polymerization process comprising combining an olefin in the gas or slurry phase with an activator, a support and a compound represented by formula (I): wherein M is a group 3 to 14 metal, each X is independently an anionic leaving group, n is the oxidation state of M, m is the formal charge of the YZL ligand, Y is a group 15 element, Z is a group 15 element, L is a group 15 or 16 element, R1 and R2 are independently a C1 to C20 hydrocarbon group, a heteroatom containing group, silicon, germanium, tin, lead, phosphorus, a halogen, R1 and R2 may also be interconnected to each other, R3 is absent, or is hydrogen, a group 14 atom containing group, a halogen, a heteroatom containing group, R4 and R5 are independently an aryl group, a substituted aryl group, a cyclic alkyl group, a substituted cyclic alkyl group, or multiple ring system, R6 and R7 are independently absent or hydrogen, halogen, a heteroatom or a hydrocarbyl group, or heteroatom containing group. Ethylene and hexene were copolymd. using a catalyst system containing [[iPrN(o-C6H4)]2O]ZrCl2·C7H8 and Me aluminoxane.

L4 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2004 ACS on STN

AN 2000:790856 CAPLUS

DN 134:100963

TI Preparation and Activation of Complexes of the Type [((mesityl)NCH2CH2)2NX]ZrMe2 (X = H, Me) with [Ph3C][B(C6F5)4] or [PhNMe2H][B(C6F5)4]

AU Schrock, Richard R.; Casado, Arturo L.; Goodman, Jonathan T.; Liang, Lan-Chang; Bonitatebus, Peter J., Jr.; Davis, William M.

CS Department of Chemistry, Massachusetts Institute of Technology, Cambridge, MA, 02139, USA

SO Organometallics (2000), 19(25), 5325-5341
CODEN: ORGND7; ISSN: 0276-7333

PB American Chemical Society

DT Journal

LA English

OS CASREACT 134:100963

AB The zirconium di-Me complexes [N2NX]ZrMe2 (N2NX = [(MesNCH2CH2)2NX]; Mes = mesityl; X = H (1a), Me (1b)), have "mer" structures in the solid state in which the amido nitrogens occupy "axial" positions in a trigonal bipyramid. The reaction of 1b with 1 equivalent of [Ph3C][B(C6F5)4] followed by addition of di-Et ether yielded the ether adduct {[N2NMe]ZrMe(Et2O)}+ (with [B(C6F5)4]- as the anion), an x-ray study of which revealed it to be a fac trigonal-bipyramidal species in which the di-Et ether is coordinated in an apical position. The reaction of 1b with 1 equivalent of [PhNMe2H][B(C6F5)4] led to {[N2NMe]ZrMe(NMe2Ph)}[B(C6F5)4], solution NMR studies of which suggest a structure analogous to that of {[N2NMe]ZrMe(Et2O)}+. Heating solns. of {[N2NMe]ZrMe(NMe2Ph)}[B(C6F5)4] led to C-H activation in one mesityl o-Me group and formation of methane. The reaction of 1b with 0.5 equiv of [Ph3C][B(C6F5)4] yielded [{[N2NMe]ZrMe}2(μ-Me)][B(C6F5)4] (5b), an x-ray diffraction study of which revealed an almost linear (167.4°) Me bridge linking two distorted TBP moieties through the apical positions (average Zr-C(bridge) = 2.48 Å, average Zr-C(terminal) = 2.24 Å). The equatorial Me groups in 5b exchange readily between Zr centers, while the bridging Me group and the equatorial Me groups exchange relatively slowly on the NMR time scale, but still rapidly on the chemical time scale. Exchange of free [N2NMe]ZrMe2 with the [N2NMe]ZrMe2 fragment in 5b is also facile on the chemical time

scale. The reaction of 1b with 1.0 equiv or more of [Ph3C][B(C6F5)4] led to formation of a cationic species (6b), two forms of which could be observed at low temperature Activation of 1a with [Ph3C][B(C6F5)4] yielded only one cationic form of 6a at low temps. Exchange of Me groups between 6a and 6b is slow on the chemical time scale. All evidence is consistent with the observation of different ion pairs of 6b at low temps.

RE.CNT 68 THERE ARE 68 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L4 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2004 ACS on STN
AN 1999:346522 CAPLUS
DN 131:144898
TI Synthesis of Group 4 Complexes that Contain the Diamidoamine Ligands, [(2,4,6-Me3C6H2NCH2CH2)2NR]2- ([Mes2N2NR]2-; R = H or CH3), and Polymerization of 1-Hexene by Activated [Mes2N2NR]ZrMe2 Complexes
AU Liang, Lan-Chang; Schrock, Richard R.; Davis, William M.; McConville, David H.
CS Department of Chemistry, Massachusetts Institute of Technology, Cambridge, MA, 02139, USA
SO Journal of the American Chemical Society (1999), 121(24), 5797-5798
CODEN: JACSAT; ISSN: 0002-7863
PB American Chemical Society
DT Journal
LA English
AB Zr complexes that contain the [(2,4,6-Me3C6H2NCH2CH2)2NR]2- ([Mes2N2NR]2-; R = H or Me) ligand, along with [Mes2N2NH]TiMe2 and [Mes2N2NH]HfMe2 complexes were prepared The activation of the Zr di-Me complexes by [Ph3C][B(C6F5)4] for the polymerization of 1-hexene at temps. up to 30° is also reported. The crystal and mol. structures of [Mes2N2NMe]ZrMe2 were determined by x-ray crystallog.

RE.CNT 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR THIS RECORD
ALL CITATIONS AVAILABLE IN THE RE FORMAT

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